

Statistical Mechanics WS 2013/14 Sheet 11

Please hand in your solutions (in pairs) before the Monday lecture.

Problem 1 : Two-Particle System (20 points)

Consider a system of a pair of two particles which are identical, and can have any of energy levels $\epsilon_n = n\epsilon$ with $n = 0, 1, 2$. The degeneracy of the lowest energy state $\epsilon_0 = 0$ is 2. Calculate the partition function and average energy for;

- (5 points) Bosons.
- (5 points) Fermions.
- (5 points) Distinguishable Classical particles.
- (5 points) Depict all the possible particle configurations in energy levels for a), b), and c).

Problem 2 : Bose-Einstein Condensation in Low Dimensions (20 points)

Consider an ideal gas of non-relativistic, identical bosons.

- (10 points) Express the mean number of bosons in one-dimension with a system length L . Explain whether there occurs the Bose-Einstein condensation or not.
- (10 points) Express the mean number of bosons in two-dimension with a system area L^2 . Explain whether there occurs the Bose-Einstein condensation or not.

[Hint: The condition for the Bose-Einstein condensation is zero-chemical potential $\mu = 0$ and low temperature $T \rightarrow 0$]

Problem 3 : Photon Gas (20 points)

Consider a photon gas in a volume V at temperature T . The energy of the photon is $\epsilon = pc$ (or $= \hbar\omega$), where p is the momentum and ω is the frequency.

- (4 points) Using canonical ensemble, calculate the average energy. Use $\epsilon = pc$.
- (4 points) Does the above result obey the “equipartition theorem”? If not, explain what is the condition for the Hamiltonian to satisfy the equipartition theorem.
- (4 points) Find the chemical potential of the photon.
- (4 points) Find the integral form for the average number of photons. How does this depend on T ?
- (4 points) Find the integral form for the average energy of photons. How does this depend on T ?

[Hint: For d) and e) use $\epsilon = \hbar\omega$ and make the integrals dimensionless.]

Problem 4 : The Planck Distribution (20 points)

The Planck distribution is photon’s radiated power R per unit area per unit wavelength λ :

$$\frac{dR}{d\lambda} = \frac{2\pi\hbar c^2}{\lambda^5 (e^{\hbar c/(\lambda k_B T)} - 1)}. \quad (1)$$

- (10 points) Using the above distribution, express the number of photons per unit area per second per unit wavelength. [Hint: Energy per photon is $\hbar c/\lambda$.]
- (5 points) Using the sun’s temperature $T = 6000$ K and wavelength range of the visible light $400 \text{ nm} \leq \lambda \leq 700 \text{ nm}$, Calculate the number of photons radiated from the sun per second per unit area [meter²].
- (5 points) What is number of photons radiated from the “cosmic background” per second per unit area [meter²]? Use $T = 2.7$ K.